KADYROVA, V.Kh.; KIRPICHNIKOV, P.A.; TOKAREVA, L.G.

Synthesis of organophosphorus stabilizers of polymers. Trudy
(KRHTI no.30:58-62 '62.

(HIRA 16:10)

MIKHAYLOV, N.V.; TOKAREVA, L.G.; POPOV, A.G.

Thermostabilization of polypropylene and fibers based on it.

Vysokomnoed. 5 no.2:188-194 F '63. (MIRA 16:2)

l. Vsesoyuznyy nauchno-issledovateliskiy institut iskusstvennogo volokna.

(Propene) (Textile fibers, Synthetic-Thermal properties)

TOKAREVA, L.G.; MIKHAYLOV, N.V.; ROZOVA, N.N.; KIRPICHNIKOV, P.A.

Lightfastness of polypropylene and fiber based on it. Khim. volok. no.3:23-25 '62. (MIRA 16:2)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna.
(Propene) (Textile fibers, Synthetic) (Photochemistry)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

VLASOV, A.V.; GLAZUNOV, P.Ya.; MIKHAYLOV, N.V.; RAFIKOV, S.R.;

TOKAHEVA, L.G.; TSETLIN, B.L.; SHABLYGIN, M.V.

Formation of oriented structures in the radiation polymerization of vinyl monomers on fibers. Dokl.AN SSSR 144 no.2:382-383 My (MIRA 15:5)

1. Institut elementoorganicheskikh soyedineniy AN SSSR i Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokhna. Predstavleno akademikom V.A.Karginym. (Vinyl compound polymers) (Radiation)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

15.800

B/190/63/005/002/005/02/ B101/B102

AUTHORS:

Mikhaylov, N. V., Tokareva, L. C., Popov, A. G.

TITLE:

Stabilization of polypropylene and of fibers made

thereof against heat

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, v. 5, no. 2, 1963,

188-194

TEXT: The effects due to 0.03 mole/kg additions of stabilizers were compared by measuring the oxygen absorption of the polypropylene at 200, 250, and 300°C and by determining the effect of the stabilizers on the breaking length of fibers drawn from the polymer at 220-250°C. At 200°C, oxidation of the polymer set in without stabilizer after an induction period of 5-7 min. The induction period was prolonged by 2,5-di-tert-butyl-4-methyl phenol (ionol) to 20 min, by T-24 (P-24) phenol - styrene copolymer to 40 min, by 2,2'-methylene-bis-(4-methyl-6-tert-butyl phenol) (2246) to 120 min and by N,N'-phenyl-cyclohexyl-p-phenylene diamine (4010) to 130 min. At 250°C a two-stage induction period was observed, particularly in the presence of dibenzyl sulfide. The first induction period was Card 1/3

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

S/190/63/005/002/005/024 B101/B102

Stabilization of polypropylene ...

10 min, the second ~ 300 min. At 300° C, the absorption curves became complicated in consequence of simultaneous thermooxidation and thermal degradation. The effects of the stabilizers on the polymer and on the drawn polymer fiber were divergent. At 200°C, and with the addition of 2246 or phenol croton aldehyde condensation product W-26 (P-26), the induction periods were respectively 120 and 130 min for the polymer, but only 45 and 80 min for the fiber. With N,N'-di- β -naphthyl-p-phenylene diamine, the induction period of the polymer was 10 min, that of the fiber 120 min. Crosslinking, and increased solubility of the stabilizer in the fiber as a result of the drawing, are suggested as explanations of the longer induction period of the fiber compared with the polymer. Reduction of the induction period can be due to the stabilizer becoming insoluble in the fiber or being decomposed in the drawing. This problem calls for further investigation. The effect of the stabilizer on the breaking length (km) and elongation (%) of the fiber after 8 hrs heating at 150°C was studied. The best results were obtained with 2,6-di-tert-butyl-4methyl-phenyl pyrocatechol phosphite, 2,6-diisobornyl-4-methyl phenol (264) 2,2'-thio-bis-(6-tert-butyl-4-methyl phenol) (KAO-6 [KAO-6]), 2264 and mixtures of stabilizers with sulfur-containing organic compounds. Without stabilizer the polypropylene fiber did not endure the test; with the Card 2/3

Stabilization of polypropylene ...

8/190/63/005/002/005/024 B101/B102

stabilizers mentioned, a residual breaking length of 60-70% was reached. No connection was found between the length of the induction period and the stabilization against heat. A synergetic effect was observed in mixtures, e.g., of 264 + 4010, ratio 1:1 (residual breaking length 67.2%). Particularly, 2,6-di-tert-butyl-4-methyl-phenyl pyrocatechol phosphite, terpene phenols and mixtures of these substances with sulfur- as well as phosphorus-containing compounds are efficient stabilizers of the polypropylene fiber. There are 1 figure and 2 tables.

-ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokna (All-Union Scientific Research Institute of Synthetic Fibers)

SUBMITTED:

July 15, 1961

Card 3/3

TOKAREVA, L.G.; MIKHAYLOV, N.V.; POTEMKINA, Z.I.; KOVALEVA, M.V.; EORIK, A.G.; ZEMSKOVA, C.N.; ZCTOVA, Ya.B.

Stabilization of polyamide fibers. Khim.volok. no.3:15-21 '61. (MIRA 14:6)

1. Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokna (for Tokareva, Mikhaylov, Potemkina, Kovaleva). 2. Klinskiy kombinat (for Borik, Zemskova). 3. Mytishchinskiy zavod (for Zotova).

(Textile fibers, Synthetic)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

S/844/62/000/000/100/129 D204/D307

Mikhaylov, N. V., <u>Tokareva, L. G.</u>, Bratchenko, T. D., Karpov, V. L. amd Malinskiy, Yu. M. AUTHORS:

The action of fradiation on artificial fibers TITLE:

Trudy II Vsesoyuznogo soveshchaniya po radiatsionnoy khimii. Ed. by L. S. Polak. Moscow, Izd-vo AN SSSR, 1962, SOURCE:

589-595

The effects of 0.05 - 1000 Mrad doses on polyamide and poly-TEXT: ester fibers, and the possibility of improving the thermal stability of synthetic fibers and improving their adhesion to rubber by the addition of various monomers, were investigated. Polyethylene terephthalic fiber was practically unaffected under doses of up to 100 Mrad, owing to the stabilizing effect of the aromatic groups, whilst a caprone fiber was already affected at 1 lirad. The specific viscosity (η) of 0.5% solutions of irradiated caprone filaments and single fibers (diameter respectively 0.03 and 0.7 mm) was meaningle fibers. sured. For the thinner fiber, η increased in vacuum and decreased

Card 1/3

S/844/62/000/000/100/129 D204/D307

٨

The action of fradiation

in air, whilst η of the monofiber increased when the latter was irradiated both in the presence and absence of air. This, and the changes in the strength and elongation showed that polyamide fibers undergo oxidative processes on irradiation; the greater changes in the presence of 0_2 were particularly pronounced for the thinner fi-

bers. Thin fibers underwent destruction when irradiated in air, whilst thicker specimens became structurized owing to the less ready diffusion of 0_2 into the mass; structurization of the thicker

fibers was also observed in vacuum. In contrast to the caprone fiber which was mainly structurized in both amorphous and crystalline states on irradiation, a terylene fiber was largely destroyed in the amorphous and structurized in the crystalline state. This difference in the behavior of polyamide and polyester fibers is ascribed to the considerably higher crystallinity of the latter. The above phenomena should be kept in mind when artificial fiber materials are to be utilized in practice. The effects of additions of acrylonitrile, styrene, toluyldiisocyanate, hexamethylenediisocyanate and vinylpyridine to the caprone fiber were studied, with

1

Card 2/3

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

1 .

The action of γ radiation ...

S/844/62/000/000/100/129 D204/D307

doses of 0.01 - 50 Mrad, finding that in all cases, for a dose of 50 Mrad, the loss in strength was considerably reduced by the monoprone fiber. There are 3 figures and 4 tables.

ASSOCIATION:

Vsesoyuznyy nacuno-issledovatel skiy institut iskusstvennykh volokon; Fiziko-khimicheskiy institut im. L. Ya. Karpova (All-Unión Scientific Research Institute of Artificial Fibers; Physico-Chemical Institute im. L. Ya. Karpov)

Card 3/3

39849

0/190/62/004/008/008/016 B101/B180

5,3833

AUTHORS:

Mikhaylov, N, V., Tokareva, L. G., Buravchenko, K. K.,

Terekhova, G. M., Kirpichnikov, P. A.

TITLE:

Stabilization of polyethylene terephthalate melts

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, v. 4, no. 8, 1962,

1186-1192

TEXT: In this fifth report on the ageing of synthetic fibers, the authors studied the thermooxidative decomposition of polyethylene terephthalate (PET) (initial intrinsic viscosity [M] = 0.245; after

reprecipitation $[\eta] = 0.256$) at 170 - 220°C (methods see Kolloidn. zh., 16, 578, 1956) and their inhibition by esters of phosphorous acid.

Results: (1) Heating to 220°C in N₂ shows no change in $[\eta]$. When heated in air, $[\eta]$ decreased more slowly in PET with reprecipitation refining than without. It is therefore assumed that thermooxidative processes occur. with the formation of COH and COOH groups and destruction of the ester bond. The PET fiber Lavsan behaved similarly: initial breaking Card 1/3

s/190/62/004/008/008/016 B101/B180

Stabilization of polyethylene ...

strength (kg/mm^2) = 42.8; after 4 hrs at 170°C in N₂, 41.0 and at 210°C in N₂, 36.5; after 4 hrs at 170°C in air, 39.5, and at 210°C, 14.2. (2) Triphenyl phosphite (I), tri-p-octyl-phenyl phosphite (II), tri-p-dodecyl phenyl phosphite (III), and tri-p-ter-butyl phenyl phosphite (IV) inhibit the thermal decomposition of PET, end increase its molecular weight and stability. The best moment for adding the inhibitor is at 50-70% polycondensation of PET. (3) After 2 hrs at 220°C the breaking strength of PET without inhibitor was 47% the initial value 71% with I, 66% with II, 78% with III, and 75% with IV. The longest induction period and smallest loss in molecular reight were found with The inhibiting effect of phosphites is attributed to the fact that they hydrolyze much more easily than PET which is thus protected against hydrolysis. There are 6 figures and 4 tables. The most important English-language reference is: J. M. Ward, Nature, 80, 141, 142, 1957.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokna (All-Union Scientific Research Institute of Synthetic Fibers)

card 2/3

Stabilizati	on of polyethylene	S/190/62/004/008/008/016 B101/E180	
SUBMITTED:	May 8, 1961	,	
•	5		
• .			:
•		•	: <i>Y</i>
	•		· V
		•	
•		•	1
ard 3/3			i

.3 TOKAREVA, L.G. 38110 3/020/62/144/002/023/028 B101/B110 15.5540 Vlasov, A. V., Glazunov, P. Ya., Mikhaylov, N. V., Rafikov, S. R., Tokareva, L. G., Tsetlin, B. L., and Shablygin, M. V. AUTHORS: Formation of oriented structures in radiation-induced polymerization of vinyl monomers on fibers TITLE: PERIODICAL: Akademiya nauk SSSR. Doklady, v. 144, no. 2, 1962, 382 - 383 TEXT: An attempt was made to obtain oriented polymers by polymerizing the monomer from the gas phase on oriented macromolecules of fibers acting as "matrices". The experiments were made with a two-chamber apparatus as used for graft polymerization of vinyl monomers on mineral particles (cf. B. L. Tsetlin et al., Tr. 2-go Vsesoyuza. soveshch. po radiatsionnoy khimii, Izd. AN SSSR, 1962). One chamber contained caprone cord fiber heated to 80°C, and the other contained completely anhydrous acrylonitrile (40°C). Irradiation was made with X-rays (dose rate, 3.1015 ev/cm3.sec) for 3 - 6 hrs at 10^{-4} - 10^{-5} mm Hg. The weight of the fiber increased by 15 - 33 %. perpendicular dichroism in the -C=N stretching vibrations (2235 cm-1), Card 1/2

S/020/62/144/002/023/028 B101/B110 3

Formation of oriented structures in ...

detected by spectroscopy, proved the orientation of the polymer. Experiments with acrylonitrile and non-oriented fiber as well as with liquid acrylonitrile and oriented fiber showed no dichroism. The liquid monomer colecules are assumed to prevent orientation. Further experiments with polymers, man-made and natural fibers used as "matrices" are under way. There is 1 figure.

ASSOCIATION: Institut elementoorganicheekikh soyedineniy Akademii nauk

SSSR (Institute of Elemental Organic Compounds of the Academy of Sciences USSR). Vsesoyuznyy nauchno-

Academy of Sciences coshy. Issuestvennogo volokna (All-Union

Scientific Research Institute of Synthetic Pibers)

PRESENTED: January 19, 1962, by V. A. Kargin, Academician

SUBMITTED: January 12, 1962

Card 2/2

MIKHAYLOV, N.V.; TOKAREVA, J.G.; KOVALEVA, M.V.

Mechanism of the aging of synthetic fibers. Part 1: Thermal and thermodridative reactions of polyamides and their fibers. Vysokom. soed. 2 no.4:581-589 Ap '60. (MIRA 13:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna.

(Polyamides) (Textile fibers, Synthetic)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

TOKAREVA, L.G.; MIKHAYLOV, N.V.; POTEMKINA, Z.I.; KOVALEVA, M.V.

mergramma made persone besiding sensebilikan kan kan kensebi

Processes and mechanism of the aging of synthetic fibers.
Part 2: Studies on the stabilization of polyamide fibers.
Vysokom. soed. 2 no. 11:1728-1738 N '60. (MIRA 13:11)

1. Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokna.

(Polyamides)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

S/190/60/002/011/023/027 B004/B060

15.8107

AUTHORS:

Tokareva, L. G., Mikhaylov, N. V., Potemkina, Z. I.,

Kovaleva, M. V.

TITLE:

Processes and Mechanism of the Aging of Synthetic Fibers.

II. Studies in the Field of Polyamide Fiber Stabilization

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2, No. 11;

pp. 1728 - 1738

TEXT: The authors have earlier studied the action of heat and light upon polyamide fibers (Ref. 3). They have arrived at the conclusion that heat and light effect irreversible oxidation processes, so that the use of antioxidants can prevent these processes from taking place. In the article under consideration, the authors deal with the action of the following antioxidants upon the stability of the caprone fiber which was heated to 200°C for two hours: N,N'-di- β -naphthyl- β -phenylene diamine; 2,2'-methylene-bis-4-methyl-6-tert-butyl phenol; 2,6-di-tert-butyl-4-methyl phenol; 2,4,6-tri-(tert-butyl)-phenol; "Poligard": [R-C₆H₄-0] 3P;

Card 1/6

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

Processes and Mechanism of the Aging of Synthetic Fibers. II. Studies in the Field of Polyamide Fiber Stabilization

Card 2/6

S/190/60/002/011/023/027 B004/B060

dimethyl phenyl-p-cresol; dibutyl-dihydroxy-diphenyl sulfide; product of the reaction of acetone with diphenyl amine; N,N°-phenyl-cyclohexyl-pphenylene diamine; 2,5-di-tert-butyl hydroquinone; product of the reaction of acetone with aminophenol; NoN'-diphenyl-p-phenylene diamine; product of the condensation of phenol with styrene, and phenyl- β -naphthyl amine. Stabilizers were added to the polymer in amounts of 0.1 to 1% prior to spinning of fiber No. 300. Additions of luminophores, such as hydroxy phenyl benzoxazole, which serve as inhibitors of the destructive action of light, indicated that these substances had a thermostabilizing effect as well. The most reliable stabilizer is said to be N, N'-di-β-naphthyl-p-phenylene diamine (DNPDA), which was used in the further experiments. Table 3 shows the action of various additions of DNPDA upon the properties of the caprone fiber. When the fiber was irradiated with a mercury lamp for 20 hours, a protective action was found to come both from DNPDA and from the luminophore hydroxy phenyl benzoxazole. The following results were obtained: 1) Aromatic diamines and their derivatives are efficient stabilizers. 2) On a long action of high temperatures upon the fiber (150°C during 100-150 h) the

Processes and Mechanism of the Aging of S/190/60/002/011/023/027 Synthetic Fibers. II. Studies in the Field B004/B060 of Polyamide Fiber Stabilization

DNPDA-stabilized fiber retained 80-85% of its original stability, while a corresponding value of no more than 20-25% was found for untreated fibers. 3) A brief action of high temperatures upon untreated fibers in nitrogen atmosphere (in the case of DNPDA-treated fibers also in the air) effects reversible changes in stability. Under these circumstances, an untreated fiber undergoes irreversible oxidative processes in the air. 4) Both thermostabilizing and photostabilizing substances exhibited the same protection both against heat and light. N. N. Semenov is mentioned. Gratitude is expressed to A. I. Korolev and his collaborators at the NIOPik (Scientific Research Institute of Organic Semifinished Materials and Dyes) for their synthesis of DNPDA, and to N. V. Demina jointly with the collaborators of the laboratoriya tekstilinykh ispytaniy (Textile Test Laboratory) for their fiber analyses.

A. M. Glebova took part in the work. There are 6 figures, 5 tables, and 6 references: 4 Soviet, 2 US, 1 British, and 1 Czechoslovakian.

Card 3/6

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

85424

Processes and Mechanism of the Aging of Synthetic Fibers. II. Studies in the Field S/190/60/002/011/023/027

B004/B060

of Polyamide Fiber Stabilization

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut

iskusstvennogo volokna (All-Union Scientific Research

Institute of Synthetic Fibers)

SUBMITTED:

July 14, 1960

Card 4/6

"APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9

•								1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				The same was the fall the same and	
موايد والموايد	e andre asses			, garagement de Cen	· · ·				в/1 воо	90/60/ 4/B060	002/01	1/023/027	To a company of the c
ī	Zot	iporpesa	бп	porpero 8 че npu 150°	1C.	9 пр	orpero 48 u npn 150°	ac.	10,110	orpero 100 · npu 150°	inc.		gran care care care
побавина	3	4 5.	3	иДочность, ХГ/жиз	i Duo,	3	профіость к17ли	удли- полио. %	3	инфиюсть	, /*		•
Количество	ж тивиц жили жили жили жили жили жили жили жи	прочность, кГ/жм² удлинение,	луд	сохране-	сохране- пие удли-		сохране-	сокране- пие упли- нения, %	пуд	сохране-	сохране- вие удли- невия, %		1
0	0,775 0,796 0,749	75,0 18,0 76,0 13,0 76,1 15,0	6 0,859 0 0,825	44.7 59.7 75.0 99.9 74.5	15,1 83,8 21,9 161 20,5	0,333 0,842 0,827	25,1 33,5 65,2 85,8 65,5 86,2	10.0 55.6 16.4 120 17.4 116 18.4	0,304 0,827 0,769	66,6 88,0 68,5 87,4	8.3 46.0 16.8 125.5 19.3 128.7 19.91		
0	0,745	75,0 17,	9 0,851	1	$\begin{array}{c c} 23.4 \\ \hline 131 \\ \end{array}$ able	0,938	82,0	18.4	0,825	83.8	111	-	X
				Ψ	ярта)							
Card	1 5/6									one of the second			

 en en e		1
	8/190/60/002/Q11/023/027 B004/B060	
Legend to Table 3:	Changes of caprone fiber properties on long heating 1 - Addition of DNPDA, %, 2 - prior to heating, 3 - 7 spec, 4 - strength, kg/mm ² , 5 - elongation, %, 6 - 8-h heating to 150°C, 7 - residual strength, %, 8 - residual elongation, %, 9 - 48-h heating to 150°C, 10 - 100 h heating to 150°C.	X
	· · · · · · · · · · · · · · · · · · ·	
		",

15.5540

2209, 1526 only

8h517 S/190/60/002/004/019/020 B004/B056

AUTHORS:

Mikhaylov, N. V., Tokareva, L. C., Kovaleva, M. V.

TITLE:

Investigation of the Mechanism of the Aging of Synthetic Fibers. I. Investigation of the Thermal and Thermal-oxidative Action Upon Polyamide and the Fibers Made From Such

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2, No. 4,

pp. 581-589

TEXT: The authors give a report on the investigation of the behavior of polyhexamethyleneadipinamide (PHMAA) and the anid fiber. Polycapronamide (PCA) and the caprone fiber at 160° to 220°C in a nitrogen atmosphere or in air and in oxygen. The volatile products were received in liquid oxygen. Of PHMAA and PCA both industrial samples as also such as were purified by re-precipitation were used. In the case of PCA, the viscosity rose with purely thermal treatment (in nitrogen) and fell as a result of thermal-oxidative treatment (Table 1). As shown by Fig. 1, the reprecipitated PCA was more stable than the non-purified substance. PHMAA

Card 1/3

Investigation of the Mechanism of the Aging of Synthetic Fibers. I. Investigation of the Thermal and Thermal-oxidative Action Upon Polyamide and the Fibers Made From Such

84517 S/190/60/002/004/019/020 B004/B056

heated in nitrogen showed a complex change in its viscosity between 1700 and 200°C (Fig. 2). A minimum at first occurred as the result of predominating destruction processes (Table 2), viz. in the non-purified polymer this occurred earlier than in the purified one. In the course of further heating, a maximum of viscosity was observed, which is explained by processes of structural formation. Finally, the polymer becomes insoluble. As analysis of Table 3 show, the elementary composition of the PHMAA changes little during heating. In the case of anid- and caprone fibers (Table 4, Fig. 3) increased viscosity occurs at first during heating at nitrogen current. In the case of oxidative heating the viscosity decreases. Above 200°C, the caprone fiber becomes insoluble more quickly than the anid fiber, which is more resistant to temperature influences. Figs. 4,6 show the change in the strength and deformation of the fibers between -80 and +200°C. The mechanical properties of the fibers in this temperature interval undergo several changes. A thermal amorphization was observed near the melting point of the crystalline

Card 2/3

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

Investigation of the Mechanism of the Aging of Synthetic Fibers. I. Investigation of the Thermal and Thermal-oxidative Action Upon Polyamide and the Fibers Made From Such .

5/190/60/002/004/019/020 B004/B056

phase. Figs. 5 and 7 show the change in the strength and deformation of fibers, which had been previously heated to various temperatures. In nitrogen, the breaking length increased, in air irreversible decrease of strength occurred. The behavior of the polyamid fibers is explained by structural transformations in the supermolecular secondary structure. The authors mention a paper by S. R. Rafikov and R. A. Sorokina (Ref. 6) A. M. Glebova, Technician, took part in the experiments. There are 7 figures, 4 tables, and 15 references: 3 Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut

iskusstvennogo volokna (All-Union Scientific Research

Institute of Synthetic Fibers)

SUBMITTED:

January 18, 1960

Card 3/3

KIRKO, V.V., dotsent; TOKAREVA, L.M.

Complications following puncture of Highmore's sinuses. Zhur. ush., nos. i gorl.bol. 22 no.1;66-69 Ja-F '62. (MIRA 15:5)

1. Iz kliniki bolezney ukha, gorla i nosa (zav. - zasluzhennyy deyatel' nauki prof. B.V.Yelantsev) Kazanskogo meditsinskogo instituta.

(ANTRUM--SURGERY)

TOKAREV, L.N., inzh.

Magnetic amplifiers without control windings. Vest. elektroprom.
34 no.5:49-52 My '63.

(Magnetic amplifiers)

(Magnetic amplifiers)

TOKAREV, Lev Nikolayevich; BALASHOV, O.I., red.; FREGER, D.P., red. izd-va; EELOGUROVA, I.A., tekhn. red.

[High-speed full-wave magnetic amplifier without control windings] Bystrodeistvuiushchii dvukhpoluperiodnyi magnitnyi usilitel' bez obmotok upravleniia. Leningrad, 1962. 22 p. (Leningradskii dom nauchno-tekhnicheskoi propagandy. Obmen peredovym opytom. Seriia: Pribory i elementy avtomatiki, no.17) (MIRA 16:5) (Magnetic amplifiers)

SAPRONOV, V.A.; KURPICHEVA, T.N.; TOKAREVA, L.T.; CHAVCHICH, T.A.; LEVIT, G.M.; BORODUSHKINA, KR.N.; BOGUSLAVSKIY, D.B.

Effect of some formula and technological factors on the quality of butyl rubber diaphragms for the forming and vulcanizing equipment. Kauch. i rez. 23 no.5:14-19 My '64. (MIRA 17:9)

1. Dnepropetrovskiy shinnyy zavod.

Quantitative devermination of the amorphous phase in pyroserams.

Rev. AN SSSR. Neorg. mat. 1 no.61979-985 Je 165.

(MIRA 18:8)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

GARIF YANOV, N.S.; TOKA: TVA, L.V.

Use of the paramagnetic resonance method in studying the crystallization of glasses. Fiz. tver. tela 6 no.5:1453-1456 My '64. (MIRA 17:9)

1. Fiziko-tekhnicheskiy institut Kazanskogo filiala AN SSSR.

Design of the Market State of the State of t

MOTOV, Sergey Iosifovich; TOKAREVA, M., red.; ANAPOL'SKTY, Ya., tekhn. red.

[Planning the production and financial operations of a state farm] Planirovanie proizvodstvenno-finansovoy deyatel'nosti sovkhoza. Moskva, M-vo sel'khoz. SSSR. No.2.[Financial planning on state farms] Finansovoe planirovanie v sovkhozakh.

1959. 141 p.

(State farms--Finance)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

TOKAKEVA, SHOL'TS, S.V.; LEVITIN, I.I., red.; TOKAREVA, M., red.; ANOPOL'SKIY, Ya., tekhn. red. [Principles of agricultureal statistics] Osnovy sel'skokhoziaistvennoi statistiki. Izd. 2. Pod. red. I.I. Levitina. Moskva. Pt. 1. 1957. (Agriculture--Statistics) 58.

的数据,是一个专家的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个

SHOL'TS, S.V.: LEVITIN, I.I., red.; TOKAREVA, M., red.; ANOPOL'SKIY, Ya., tekhn. red.

[Principles of agricultural statistics] Osnovy sel'skokhoziaistvennoi statistiki. Izd.2. Pod red. I.I. Levitina. Moskva. Pt.1. 1957.
58.p. (MIRA 11:10)
(Agriculture—Statistics)

IAUYUR, A.P.; TOKARNYA, M., red.; ANOPOL'SKIY, Ye., tekhn.red.

[Business correspondence for collective farms] Deloproizvodstvo v

[Business correspondence for collective farms] Deloproizvodstvo v

[kolkhozekh. Moskva, Upravlenie podgotovki kadrov M-ve sel'khoz.

(MIRA 11:2)

SSSR, 1957. 110 p.

(Gommercial correspondence) (Gollective farms)

ISHNEVSKIY, M. (Simferopol'). (MLRA 6:		
Road to mastery	. Radio no.6:13 Je '53.	(Tokareva, Mariia)

SHOL'TS, S.V.; LEVITIN, I.I., otv. red.; TOKAREVA, M., red.; YANOPOL'SKIY, Ya., tekhn. red.

[Principals of agricultural statistics] Osnovy sel'skokhoziaistvennoi statistiki. Pod red. I.I.Levitina. Moskva, Upravlenie podgotovki kadrov M-va sel'.khoz.SSSR. No.3. 1955. 41 p. (MIRA 15:4)

(Agriculture-Statistics)

ABRAMOVA, Aleksandra Afanas'yevna, dots.; KARAVAYEV, Valentin
Valentinovich, dots.; kand. yurid. nauk; TOKAREVA, M.,
red.; TRATNIKOVA, V., tekhn. red.

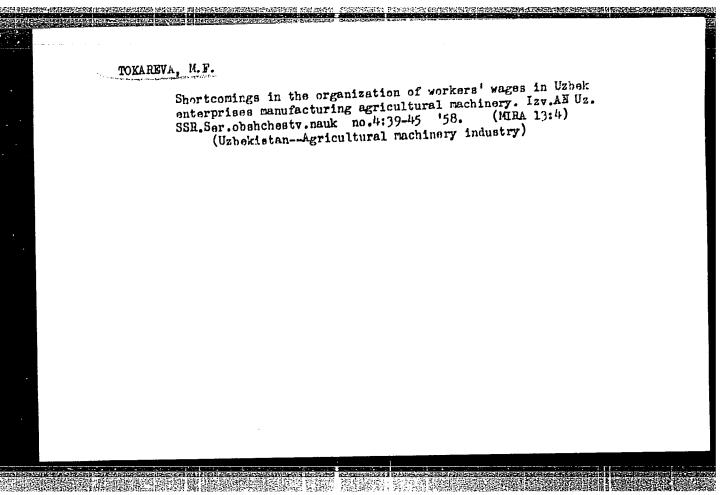
[Basic principles of Soviet labor law] Osnovy sovetskogo
trudovogo prava. Izd.3., perer. Pod red. V.V. Karavaeva.
Moskva, VZUK, 1962. 139 p.
(Labor laws and legislation)

(Labor laws and legislation)

BERGER, A.S.; TOKAREVA, M.D.

Fhasic composition of products of the interaction of \$\beta\$-dicalclum silicate with soda-aluminate solutions. Izv. Sib. otd. AV SSER no.9:71-79 162. (MIR: 17:8)

1. Khimiko-metallurgicheskiy institut Sibirskogo otdeleniya AN SSSR, Novosibirsk.



- 1. GALUSHKINA, N. A., TOKAREVA, M. F.
- 2. SSSR (600)
- 4. Uzbekistan-Oil Industries
- 7. Production capacity potentials of the Uzbekistan oil factories.
 Masl. zhir. prom. 17 No. 5, 1952

9. Monthly List of Russian Accessions, Library of Congress, February 1953, Unclassified.

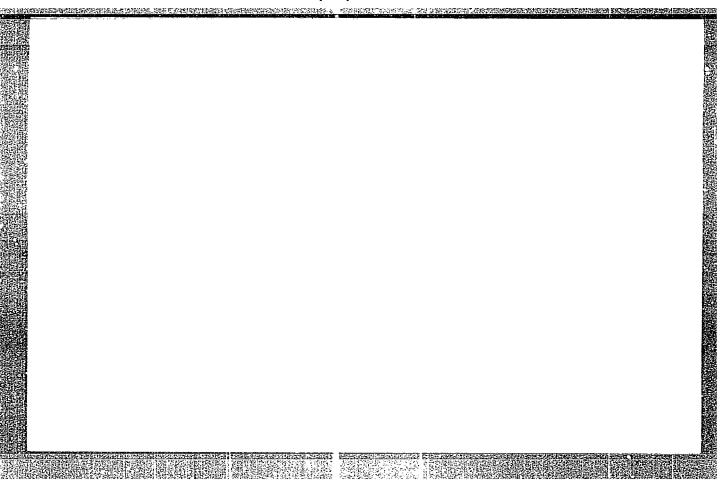
- 1. GATUSHKINA, N.A.: TOKAPEVA, M.F.
- 2. USSR (600)
- 4. Oil Industries Uzbekistan
- 7. Production capacity potentials of the Uzbekistan oil factories. Masl. zhir. prom. 17. no. 5. 1952.

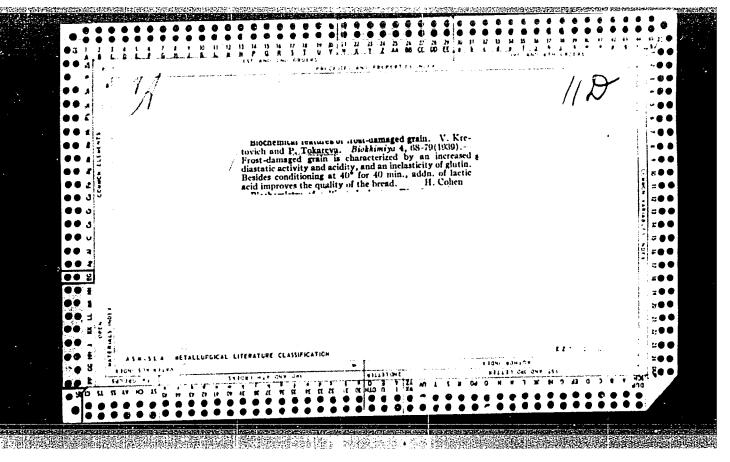
9. Monthly List of Russian Accessions, Library of Congress, February 1953, Unclassified.

TOK-AKE KIT NIL

"The Mutual System Consisting of the Chlorides and Mitrates of Lithium and Strontium," by M. V. Tokareva and A. G. Bergman, Rostov-na-Donu State University and Voroshilovgrad State Pedagogic Institute, Zhurnal Neorganicheskoy Khimii, Vol 1, No 11, Nov 56, pp 2570-2576

Results of an investigation of the system Li,Sr//Cl, NO3 are reported. These results supplement data obtained earlier on the systems Li, Sr//Cl,SO4 and Li,Sr//Cl,F. The paper represents part of a more extensive investigation in which an attempt has been made to determine the effect of different anions on the alkali metals - alkaline earth cations exchange in salt melts. On the basis of the results obtained in this instance, the conclusion is made that the system Li,Sr//Cl,NO3 is analogous to the system Li,Sr//Cl,SO4, except that it is reversible to a greater extent. It is furthermore concluded that the system Li,Sr//Cl,NO3 differs from the system Li,Sr//Cl,F in that complex-formation predominates in the latter, while cation exchange predominates in the first. Data on the binary systems $(\text{LiNO}_3)_2$ - $\text{Sr}(\text{NO}_3)_2$, SrCl_2 - $\text{Sr}(\text{NO}_3)_2$, LiNO_3 - LiCl, and Li_2Cl_2 - SrCl_2 and on the diagonal sections $(\text{Sr}(\text{NO}_3)_2 - \text{Li}_2\text{Cl}_2)$ and $(\text{LiNO}_3)_2 - \text{Sr}\text{Cl}_2$ are given. (U)





TOKAREVA, M.V.; RUDENKO, V.K.

Reciprocal system consisting of lithium and varium nitrates and chlorides. Zeur.neorg.khim. 8 no.3:702-707 Mr *63. (MIRA 16:4)

1. Luganskiy gosudarstvennyy pedagogicheskiy institut. (Systems (Chemistry))

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

AUTHORS:

Tokareva, 1. V., Bergman, A. G.,

30V/78-3-9-31/48

Kayalova, S. S.

TITLE:

Reciprocal System of Mitrates and Chlorides of Sodium and Calcium (Vzaimnaya sistema iz nitratov i khloridov natriya

i kal'tsiya)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1958, Vol. 3, Nr 8,

pp. 1909-1913 (USSR)

ABSTRACT:

Details were given of the experimental examinations of the reciprocal system of nitrates and chlorides of sodium and calcium in the presence of solvents. These examinations were performed by means of visual-polythermal methods. The system Na, Ca | Cl, NG is comparatively simple and the components

forming the system do not react with each other by forming complex compounds and solid solutions. This system is analogous to the systems: Na, Sr $\|C1$, NO₃ and Na, Ba $\|C1$, NO₃.

The results demonstrate that the reciprocal system

Na, Ca | Cl, NO, belongs to the most simple reciprocal systems.

Card 1/2

There are 9 figures, 3 tables, and 10 references, 10 of which

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

Reciprocal System of Mitrates and Chlorides of SOV/78-3-8-31/49
Sodium and Calcium

are Soviet.

ASSOCIATION: Rostovskiy-na-Donu gosudarstvennyy universitet i Luganskiy gosudarstvennyy nedngogicheskiy institut (State University Rostov na Donu and the State Pedagogical Institute, Lugansk)

SUBMITTED: July 21, 1957

Card 2/2

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

TOKAREVA, M.V.; BERGMAN, A.G.; KAYALOVA, S.S.

Reciprocal system consisting of sodium and calcium nitrates and chlorides. Zhur. neorg. khim. 3 no.8:1909-1913 Ag *58.

(MIRA 11:9)

1. Rostovskiy-na-Donu gosudarstvennyy universitet i Luganskiy gosudarstvennyy pedagogicheskiy institut.
(Systems (Chemistry))

TOKAREVA, MIVI

USSR/Physical Chemistry - Thermodynamics, Thermochemistry, B-

Equilibria, Physical-Chemical Analysis, Phase Transitions.

Abs Jour : Referat Zhur - Khimiya, No 1, 1958, 407

Author : A.G. Bergman, M.V. Tokareva.

Title : Interaction Between Silver Nitrate and Chlorides of Alkali

Earth Metal in Absence of Solvent.

Orig Pub : Zh. neorgan. khimii, 1957, 2, No 5, 1086-1093

Abstract : The system of Ag and Ca chlorides (I and II) and of Ag and

Ca nitrates (III and IV), as well as the diagonal sections III - $BaCl_2$ (V) and III - $SrCl_2$ (VI) were studied by the visual-polythermal method. The system is irreversibly reciprocal and singular. There is a shift of the metathetical reaction to the side of I - $Me(NO_3)_2$ (VII) in the earlier studied reciprocal systems of Ag, K and Li, and the most refractory component of the stable diagonal appears

as the exchange product. The study of the diagonal

Card 1/2

USSR/Physical Chemistry - Thermodynamics, Thermochemistry, B-8 Equilibria, Physical-Chemical Analysis, Phase Transitions.

Abs Jour : Ref Zhur - Khimiya, No 1, 1958, 407

systems of these salts were also singular with a sharp shift of metathesis to the side of VII - I. The character of the system Ag, Sr // Cl, NO₃ is the most singular. Numerical data and graphs of states of the systems are given.

Card 2/2

(Systems (Chemistry))

ERROMAN, A.G.; TOKAREVA, M.V.

Musibility diagram in a system of harium and calcium nitrates and chlorides. Zhur. neorg. khim. 2 no.8:1888-1894 Ag '57. (MIRA 11:3)

1. Rostovskiy-na-Donu gosudarstvennyy universitet i Voroshilov-

gradskiy gosudarstvennyy pedagogicheskiy institut.

TOKARHVA, M.V.; BERGMAN, A.G.

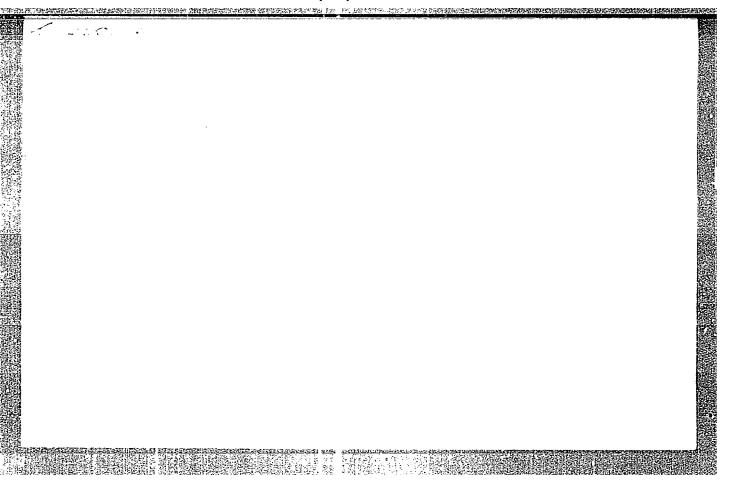
Fusibility diagram in a system of powerstand and nitrates. Zhur. neorg. khim. 2 no.8:1895-1906 Ag '57.

(MIRA 11:3) Fusibility diagram in a system of potassium and strontium chlorides

1. Rostovskiy-na-Domu gosudarstvennyy universitet i Voroshilovgradskiy gosudarstvennyy pedagogicheskiy institut.

(Systems (Chemistry))

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"



BERGMAN, A.G.; TOKAREVA, M.V.

Reaction of silver nitrate with chlorides of alkali earth metals in the absence of a solvent. Zhur. neorg. khim. 2 no.5: 1086-1093 lty '57.

1. Rostovskiy gosudarstvennyy universitet imeni V.M. Molotova i Voroshilovgradskiy pedagogicheskiy institut ineni T.G. Shevehenko.

(Silver nitrate) (Alkali metals) (Chlorides)

TOWARDWA, M. W.: "Investigation of the chloride-nitrate exchance of rone- and divalent metals in the fused state." Rostev na Bone State & Snort M. M. Moleley. Chair of General and Incorposit Chemistry. Scatev na Bone, 1096. (Na contest of for the Degree of Candidate in Chemical Science.)

Whizhnaya letopist, No. 30, 1096. Messer.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

Reciprocal system of lithium and strontium chlorides and nitrates. Zhur. neorg. khim. 1 no.11:2570-2576 H '56.

(MLRA 10:5)

1. Rostovskiy-na-Donu gosudarstvennyy universitet i Voroshilovgradskiy gosudarstvennyy pedagogicheskiy institut.

(Lithium salts) (Strontium salts) (Systems (Chemistry))

TOKAREVA, M. V., and BERGMAN, A. G.

"Mutual System Composed of the Chlorides and Nitrates of Lithium and Strontium," by M. V. Tokareva and A. G. Bergman, Rostov-na-Donu State University and Voroshilovgrad State Pedagogic Institute, Zhurnal Neorganicheskoy Khimii, Vol 1, No 11, Nov 56, pp 2570-2576

The constitutional diagram of the system Li,Sr//Cl,NO $_3$ has been investigated. The results of the work described in this instance supplement data on the systems Li,Sr//Cl,SO $_{l_4}$ and Li,Sr//Cl,F, which have been studied earlier.

Sum 1274

SHATS, V.Ya.; TOKAREVA, N.A.

Prolonged anticoagulant therapy under outpatient conditions. Kaz.med.zhur. no.5:84-86 S-0 '62. (MIRA 16:4)

FEDORIN, Yu.V.; TOKAREVA, N.P.

Agrochemical characteristics of swamp in Semirech'ye. Pochvo-vedenie no. 12:7-11 D '65 (MIRA 19:1)

1. Kazakhskiy institut zemledeliya. Submitted March 22, 1965.

KRAYEVSKIY, A.A.; ZHELEZNOVA, Ye.S.; TOKAREVA, N.V.

Obtaining the alkaloid triacanthine from the leaves of Gleditschia triacanthos L. Med.prom. 14 no.4:30-33 Ap '60. (MIRA 13:6)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut lekarstvennykh i aromaticheskikh rasteniy.

(TRIACANTHINE)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

MOROZOV, G.; SAVIN, N.; TOKAREVA, O.; KAL'NIN, O.

We lewer preduction cests. Prem.keep.ne.10:20-22 O '55,
(Efficiency, Industrial)

(MIRA 9:4)

FEDOTOV, N.P., grof. (Tomsk); TOKAREVA, O.G. (Semipalatinsk)

F.B. Gebler, prominent Siberian physician. Trudy Perm. gos. med. inst. 43:171~174 '63. (MIRA 17:6)

TOKAREVA, O. G., CAND MED SCI, "MATERIAL ON THE HISTORY OF FORENSIC MEDICAL SERVICE IN WESTERN SIBERIA IN THE PREREFORM PERIOD." SEMIPALATINSK, 1961. (TOMSK STATE MED INST).

(KL, 3-61, 236).

474

TOKAREVA, O.N.

New electronic mathematical machine for calculating frame systems.

Vych. i org.tekh. v stroi. i prock. no.1:88-89 '64.

(MIRA 18:10)

1. Institut kibernetiki AN UkrSSR.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

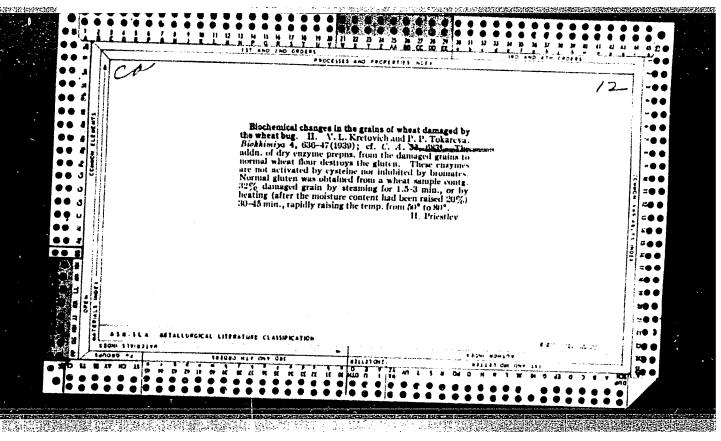
PUKHOV, Georgiy Yevgen'yevich; VASIL'YEV, Vsevolod Viktorovich;
STEPANOV, Arkadiy Yevgen'yevich; TOKAREVA, Ol'ga Nikolayevna;
BMAS, R.L., red.izd-va; RAKHLINA, N.P., tekhn. red.; REKES,
M.A., tekhn. red.

[Electric modeling of problems in structural mechanics] Elektricheskoe modelirovanie zadach stroitel'noi mekhaniki. [By] G.E.Pukhov i dr. Kiev, Izd-vo AN USSR, 1963. 285 p. (MIRA 17:3)

1. Chlen-korrespondent AN Ukr.SSR (for Pukhov).

Some problems concerning the solution of linear programming problems. Mat. mod. i elek. tsepi no.1:28-32 '63. (MIRA 16:11)

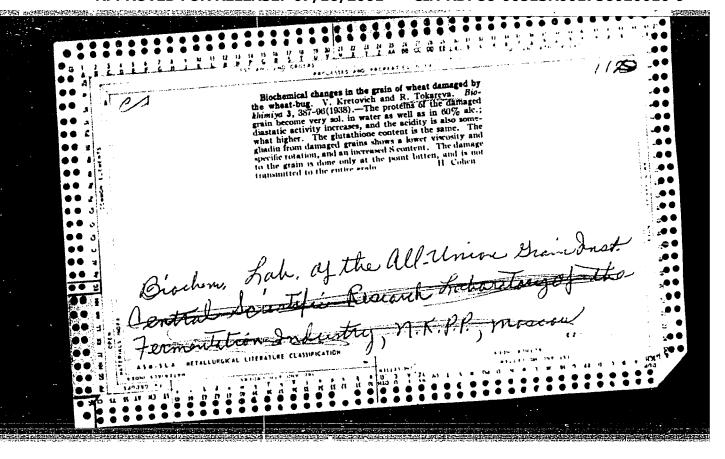
APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

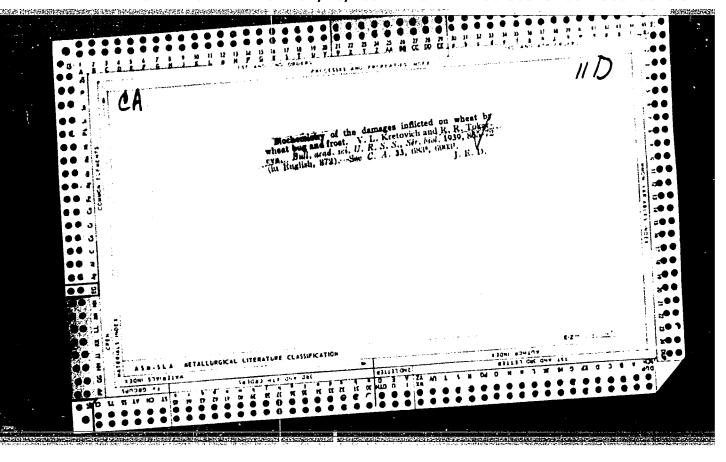


CHERKASSKAYA, A.R.; PEREL'MUTER, Ye.A.; TOKAREVA, R.O.

Pneumoencephalographic studies in organic psychoses in childhood and adolescence. Zhur. nevr. i psikh. 64 no.7: 1070-1073 '64. (MIRA 17:12)

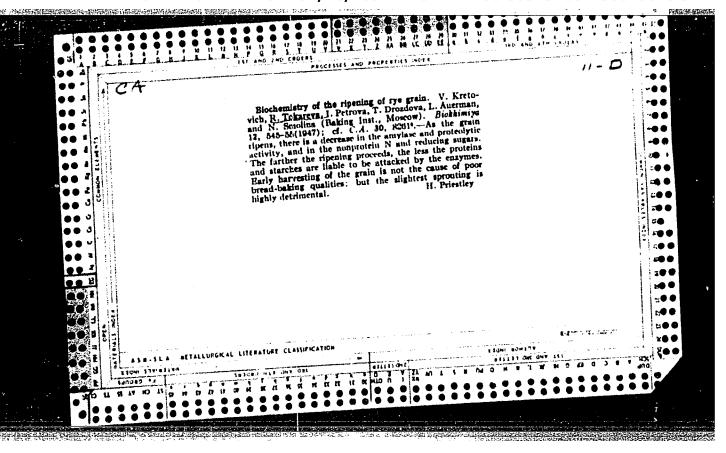
1. Kafedra psikhiatrii (zaveduyushchiy - prof. L.A. Mirel'zon) Odesskogo meditsinskogo instituta i Odesskaya oblastnaya psikhonevrologicheskaya bol'nitsa (glavnyy vrach F.K. Filyanovskiy).





TOKAREVA, R.R.

Biochemical changes in the grains of wheat damaged by wheat-bug. V.L. KRETOVICH AN D R.R. TOKAREVA. (BIOCHEMICAL LABORATORY, ALL UNION INSTITUTE OF GRAIN, MOSCOW) biokhimiya 4, no.6, p. 636, 1939.



TOKAREVA R., KRETOVICH V., AUERMAN L., SMOLINA N., KUIMAN A., BRANCFOL'SKAYA, R.

"Change in the Quality of Rye Flour During Storage," Dok. An, 58, Mc. 9, 1947.

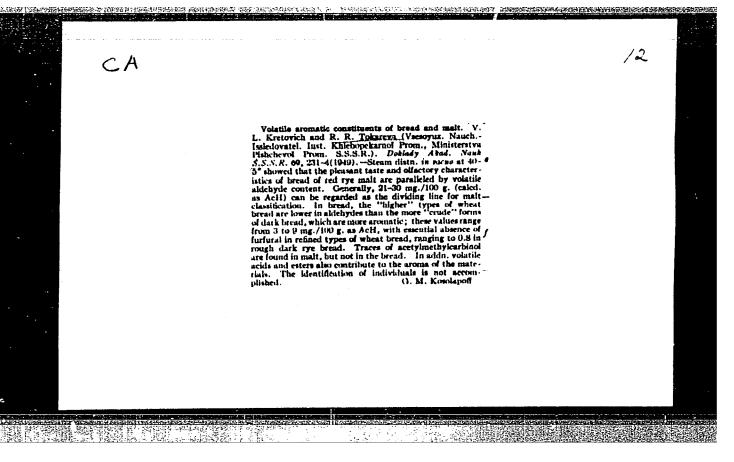
KRETOVICH, V.; TOKAREVA, R.

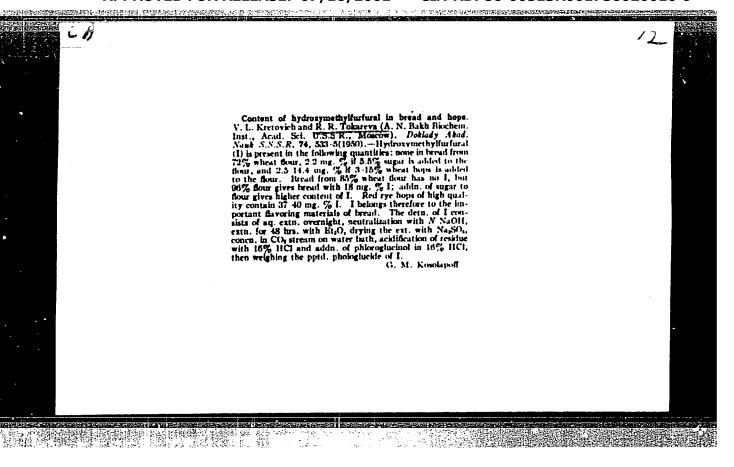
Inst. Baking Industry, -c1948-; Inst. Biochemistry im. A. N. Bakh, USSR Acad. Sci., Moscow, -c1948-.

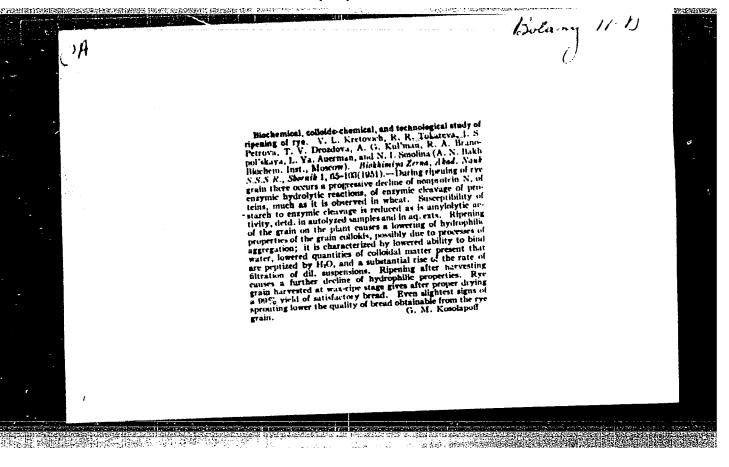
"Reaction of amino acids and sugars at high temperatures," Biomhimiya, 13: 6, 1948.

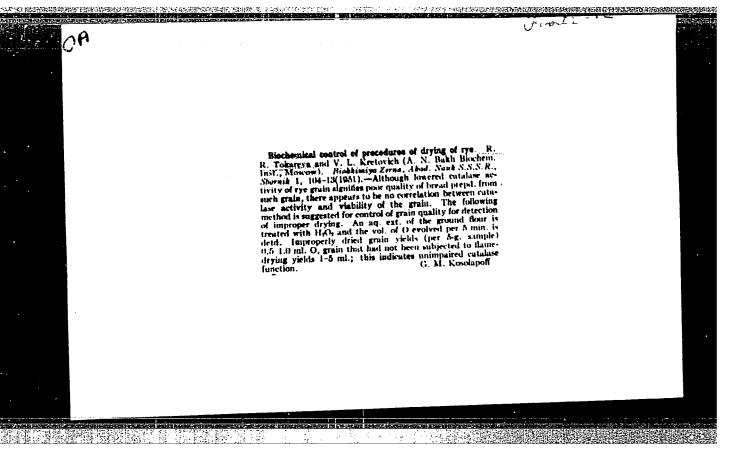
BNL Guide, 2: 4,1949.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"



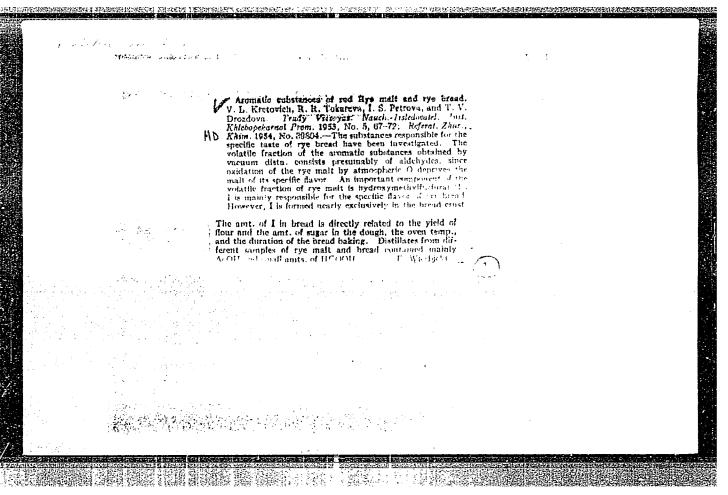






- 1. KRETOVICH, V. L.- TOKAREVA, R. R. PETROVA, I. S. DROZDOVA, T. V. KUL'MAN A. G. BRANOPOL'SKAYA, R. A. AUYERMAN, L. YA. SMOLINA, N. I.
- 2. USSR (600)
- 4. Wheat
- 7. Biochemical, colloid-chemical, and technological studies of the maturing of wheat. Biokhim.zerna no. 1, 1952

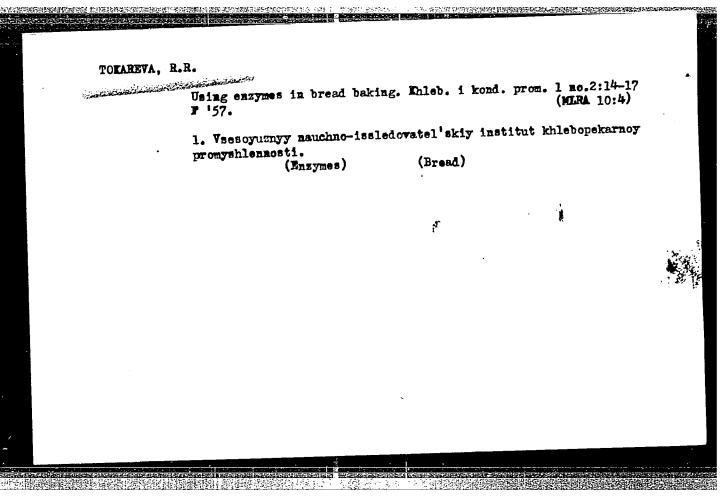
9. Monthly List of Russian Accessions, Library of Congress, March 1953, Unclassified.



TOKAREVH, K.15.

1-25-54 Yoods Melanoidin formation and color of bread crust. L. Yd. Auerman, V. L. Kretovich, E. A. Alyakrinskaya, V. M. Bazarnova, and R. R. Tokareva (A. N. Bakh Biochem. Inst. Acad. Sci. U.S.S.R., Moscow). Daklady Akad. Nauk S.S.S.R. 92, 131-3(1953).—When wheat grain is dried at elevated temp. (150°) the protein-proteinase system undergoes profound changes: water-sol. N. raw gluten content, and its H₂O-absorbing power decline, with almost complete inactivation of the proteinases. The bread baked from the flour prepd. from such grain has low porosity and high d., owing to poor gas retention. However, the crust of such bread is unusually light in color. This is explained by the lack of proteinase activity since this fact causes a lack of the necessary carbohydrate materials which act as raw materials for melanoidin formation which produces the normal crust color. When maltose, fructose, sucrose, and glycine were added to the deficient flour, the resulting bread had a more pigmented crust; glycine was particularly effective, and the full complement of glycine and one of the disaccharides gave normal color. Thus the color is produced by interaction of reducing sugars with products of protein hydrolysis.

G. M. Kosolapoff



TOKAREVA, R.R.; SMIRNOVA, G.M.; KRETOVICH, V.L.

[Use of ferment preparations in the bread making industry] Primenenie fermentnykh preparatov v khlebo-pekarnoi promyshlemosti. Moskva, TSentr. in-t nauchno-pekarnoi promyshlemosti. 1963. 68 p. tekhn. informatsii pishchevoi promyshl., 1963. 68 p. (MIRA 17:9)

SMIRNOVA, G.M.; TOKAREVA, R.R.; KRETOVICH, V.L. the second of the second of the second of

Ferment preparations as regulators of biochemical and microbiological processes in the making of rye bread.
Biokhim. zer. i khlebopech. no.7:245-263 164.

(MIRA 17:9)

1. Vsesoyuznyy nauchno-issledovatel skiy institut khlebope. karnoy promyshlennosti i Institut biokhimii imeni Bakha AN SSSR.

CIA-RDP86-00513R001756020010-9" **APPROVED FOR RELEASE: 07/16/2001**

TOKAREVA, R.R.; SMIRNOVA, G.M.; YEGOROVA, L.A.; KALININA, V.1.

Use of enzymes for improving the quality of bread Lade from 157
quality flour. Trudy TSNIIKHP no.10:138-147 '62.

(MIRA 18:2)

fune	estigating the ferment rus for the improvemen 3:135-140 '60. (Fermentati	C OT Diem dams	the new types of mo ty. Trudy TSNIIKH (MIRA	15:8)

TOKAREVA, R. R., KRETOVICH, V. L. (USSR)

"The Use of Concentrated Enzyme Preparations from Mould Fungi in Bread Making."

Report presented at the 5th Int'l. Biochemistry Congress, Moscow, 10-16 Aug 1961.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

TSENTSIFER, A.B.; TOKAREVA, S.A.

Reaction of carbon monoxide with sodium and potassium superoxides.

Zhur.neorg.khim. 6 no.11:24.74-24.80 '61. (MIRA 14:10)

(Carbon monoxide) (Sodium superoxide)

(Potassium superoxide)

KUZNETSOV, V.G.; TOKAREVA, S.A.; DOBROLYUBOVA, M.S.

X-ray diffraction study of sodium ozonide NaO3. Zhur.neorg.khim.
7 no.5:967-970 My '62. (MIRA 15:7)
(Sodium oxides) (X rays-Diffraction)

ACCESSION NR: AP4033391

5/0062/64/000/004/0739/0740

AUTHOR: Tokareva, S. A.; Dobrolyubova, M. S.

TITIE: Ozonidizing sodium hydroxide in the -100 to 50C temperature range

SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 4, 1964, 739-740

TOPIC TAGS: sodium hydroxide, ozonization sodium ozonide, NaO sub 3, stability, sodium superperoxide, synthesis

ABSTRACT: The process of ozonidizing NaOH at -100 to +50C was investigated. The possibility of forming NaO₃ depends on the experimental conditions: at a slow ozone-oxygen feed rate, the ozone decomposes; at very high rates and low temperatures, the contact time is too short. With an ozone-oxygen mixture feed of 20 liters/hour, noticeable ozonidation in NaOH occurs at -60C. At 200 liters/hour ozone-oxygen feed rate, ozonidation of NaOH was insignificant at -100 to -40C; in the -20 to +50C range a maximum yield of about 2% NaO₃ (on the weight of the initial alkali) was obtained at OC, with yield reduced to 0.18-0.2% at 50C. The material is storable for several months in the absence of atmospheric moisture. Small amounts of sodium superperoxide were formed simultaneously during ozonidation.

'Card' 1/2

ACCESSION NR: AP4033391

Orig. art. has: 3 figures.

ASSOCIATION: Institut obshchey i neorganicheskoy khimii im. N. S. Kurnakova Akademii nauk SSSR (Institute of General and Inorganic Chemistry Academy of

Sciences SSSR)

SUBMITTED: 11Mar63

ENCL:

SUB CODE: IC

NO REF SOV: 002

OTHER: 002

Card 2/2

CIA-RDP86-00513R001756020010-9" **APPROVED FOR RELEASE: 07/16/2001**

ACCESSION NEW YORK AND A SECRET AS SECURIOR OF THE SECRET AND ACCESSION OF THE SECRET

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001756020010-9"

L 16111-55

ACCESSION NR: AP4045837

were monoaxial, negative, with strong double refraction, their indices were $N_p = 1.391$, $N_g = 1.670$ The densities were found it about 1.56-1.60 gc. c. for the Na and it I want give for the E. Maride. These acondes have los so be $((A^{k+1})^{-1})^{-1} = ((A^{k+1})^{-k})^{-1} = ((A^$

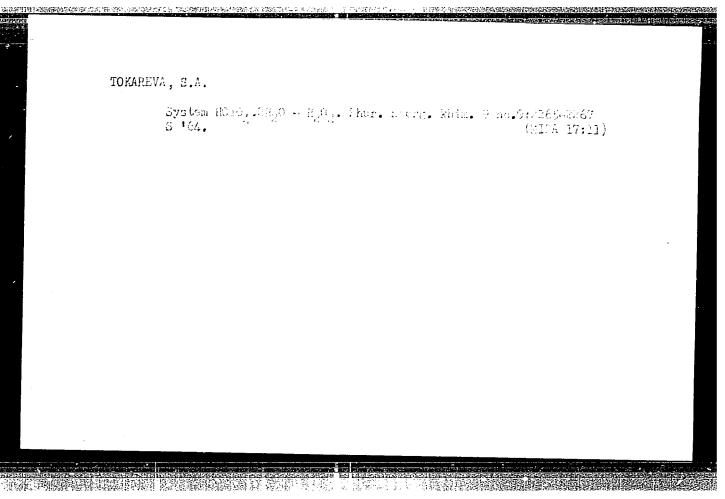
ASSOCIATION: Institut obshchey i neorganicheskov khimil im N. S. Kurnakaya We have been a sectional control of the section of the property of the section of

SUBMITTED: 20Jun63

ENCL: 00

SUB CODE: GC, IC, GP NO REL SOV: 007 OTHER 003

Card 2/2



ACCESSION NR: AT4028338

8/0000/63/000/000/0188/0192

AUTHOR: Tokareva, S. A.; Dobrolyubova, M. S.; Makarov, S. Z. (deceased)

TITLE: Study of the NaOH ozonization process at low temperatures and identification of the physical chemical properties of sodium ozonide

SOURCE: Soveshchaniye po khimii perekisny*kh soyedineniy. Second, Moscow, 1961. Khimiya perekisny*kh soyedineniy (chemistry of peroxide compounds); Doklady* soveshchaniy. Moscow, Izd-vo AN SSSR, 1963, 188-192

TOPIC TAGS: sodium hydroxide, sodium ozonide, ozonization, ozone, hydroxal, potassium hydroxide, alkali, ammonia, ozonide, sodium, potassium

ABSTRACT: In this paper the authors discuss the ozonization process of sodium hydroxide, the precipitation of sodium ozonide in a crystalline state, and the study of the properties of sodium ozonide. Crystalline sodium ozonide with a NaO₃ content of 80-90% was precipitated. The ozonization process of sodium hydroxide is studied within a temperature range of from +40 through -100°C. The formation process of sodium superoxide in the ozonization of sodium hydroxide in a temperature range of -40 through -100°C is also studied. X-ray examination of sodium ozonide was made. An x-ray analysis yielded a satisfactory agreement between the experimental

Card 1/2

ACCESSION NR: AT4028338 and calculated values of $\sin^2 0$ for a body centered tetragonal nucleus with a period of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide was of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium ozonide to the conductant of a = 11.61Å; c = 7.66Å. A differential thermal analysis of sodium o		. %			!			
SUBMITTED: 13Dec63 DATE ACQ: U6APF04 OTHER: 007 NO REF SOV: 007	and calculated of a = 11.61Å; conducted. The dissociation of 3 figures and	values of c = 7.66A e exothern E sodium o l formula	sin ² 0 for A diffical effect ozonide is	of -10 thr reproduced	on all the	rmograms. O	rig. art. ha urnakova AN S	ssr
	(Institute of	General a	<u>1</u> .				ENC	L: 00
	SUBMITTED: 13	Genera-		date acq:	06Apr64		ENC	L: 00
	SUBMITTED: 13	Genera-		date acq:	06Apr64		ENC	L: 00

TOKAREVA, S. A.; DOBROLYI BOVA, M. S.

Ozonization of sodium hydroxide within the temperature range of 100 to 50° C. Izv AN SSSR Ser Khim no. 4:739-740 Ap 164. (MIRA 17:5)

1. Institut obshchey i neorganicheskoy khimii im. N. S. Kurnakova AN SSSR.